Guavin: An Unusual Oxalactone Dibenzxanthone from Tabebuia guayacan

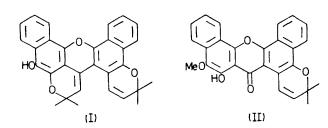
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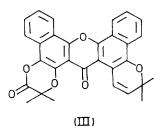
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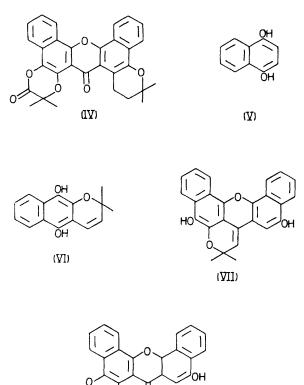
Summary A unique oxalactone dibenzxanthone has been identified in Tabebuia guayacan wood and its dihydro derivative has been successfully synthesised.

It has previously been reported¹ that the marine borer resistant heartwood of Tabebuia guayacan Hemsl. contains a novel phenolic xanthene derivative guayacanin (I), tectol, and several of the naphthaquinones found in other Tabebuia species.² A new and unusual compound, guayin, (III) m.p. 277-278 °C, has now been isolated from the ether extract of the heartwood.



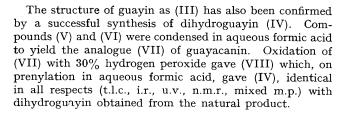


remaining carbonyl group. On the basis of this and other spectral properties this phenol was considered to be (II) and guayin is formulated as the dibenzxanthone derivative (III). X-ray crystallographic analysis of guayin has confirmed this structural assignment. The similarity of structure (I) and (III) suggest the derivation of guayin from guayacanin (I) by oxidative cleavage and subsequent rearrangement of the benzylically substituted chromene ring.



Guayin, $C_{30}H_{22}O_6$, λ_{max} (EtOH) 414 (log ϵ 3.84), 323 (4.22), ca. 294 (4.53), 282 (4.57), ca. 250 (4.51), 226 nm (4.81) is non-phenolic and contains one conjugated carbonyl (i.r. 1660 cm⁻¹) and one six-membered lactone carbonyl (i.r. 1785 cm⁻¹). The 100 MHz n.m.r. spectrum in CDCl₃ shows two geminal dimethyl groups [δ 1.56 (6H, s), 1.77 (6H, s)], two vicinal vinylic (chromene) protons [δ 5.75 (1H, d, I 10) Hz), 8.01 (1H, d, J 10 Hz)] and eight aromatic protons $[\delta 8.56 - 8.74 (8H, m)].$

The presence of the lactone function was confirmed by alkaline methylation of guayin to form the OO-dimethyl derivative of the corresponding hydroxyacid (C₃₂H₂₈O₇, m.p. 186-187 °C). Hydrolysis of the ester yielded the acid, decarboxylation of which was accompanied by the loss of a $C_4H_6O_2$ unit, yielding a product with the properties of a catechol monomethyl ether derivative, $C_{27}H_{20}O_5$. The u.v. spectrum of this phenol showed a pronounced bathochromic shift (24 nm) upon addition of aluminum chloride,³ indicating the free phenolic hydroxy group to be peri to the



(YIII)

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- ² A. R. Burnett and R. H. Thomson, J. Chem. Soc. (C), 1967, 1710; 1968, 850.
 ³ L. Jurd, in 'The Chemistry of Flavonoid Compounds', Pergamon Press, New York, 1962, p. 119.